

EXPERIMENTAL RESULTS OF $\text{La}_{0.8}\text{Sr}_{0.2}\text{CrO}_3/\text{SrTiO}_3(100)$ HETEROSTRUCTURE USED FOR A HIGHTEMPERATURE PHOTOVOLTAIC CELL

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Abstract

The absorption of light by semiconducting materials results in a partial conversion of the radiative energy into electrical energy, a process of great importance and interest in energy technology. The solar spectrum can be classified into a high energetic part (short wave length) and a lower energetic part (long wave length). Both can be used by the special type of high temperature photovoltaic cell (HT-PV) under consideration, which consists of solid oxide materials, especially perovskite oxides with the general formula ABO_3 . In fact, strontium doped lanthanum chromium oxide ($\text{La}_{1-x}\text{Sr}_x\text{CrO}_3$) and strontium titanate (SrTiO_3) are not only attractive candidates as semiconducting oxide materials for a HT-PV cell, but also for driving an electrochemical reaction used in a *Solid Oxide Photo-Electrochemical Cell* (SOPEC).

Material Characterisation

This paper shows the results of investigating the heterostructure $\text{La}_{0.8}\text{Sr}_{0.2}\text{CrO}_3/\text{SrTiO}_3(100)$ under ultra-violet (UV) radiation at temperatures from 400°C up to 500°C. The HT-PV cell consists of a SrTiO_3 (STO) single crystal in the orientation 100 with a $\text{La}_{0.8}\text{Sr}_{0.2}\text{CrO}_3$ thin-film deposited by *pulsed laser deposition* (PLD). Both perovskites are large band gap (E_g) semiconducting metal oxides. It is well known that the E_g of STO is about 3.2 eV. Furthermore, optical measurements of 20% doped strontium LaCrO_3 ($\text{La}_{0.8}\text{Sr}_{0.2}\text{CrO}_3$) have shown that its E_g should be around 2.8 – 3.1 eV. For the excitation of electron-hole charge carriers (e^- , h^+) photons in wave-length ν equal to or larger than around 385 nm are necessary. Any photon's energy that is higher or lower than E_g is converted into heat. Given the high E_g of the $\text{La}_{0.8}\text{Sr}_{0.2}\text{CrO}_3/\text{SrTiO}_3(100)$ -system, electrons (e^-) can be excited from the valence band (VB) to the conduction band (CB) at temperatures up to 600°C. In contrast, e^- become thermally excited at relatively low temperatures when using low-band gap materials such as Si-based semiconductors (~1.1eV).

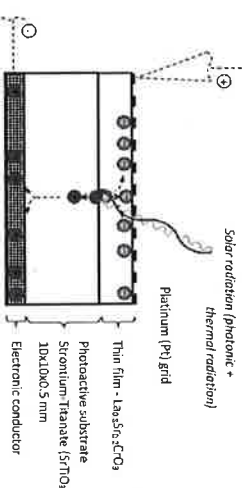


Fig. 1. Scheme of the heterostructure $\text{La}_{0.8}\text{Sr}_{0.2}\text{CrO}_3/\text{SrTiO}_3(100)$ single crystal used as HT-PV cell

Research Process and Results

In a first step, the HT-PV cells were characterized by measuring current-voltage (I-V) curves. The I-V measurement was performed depending on a) temperature, b) light intensity and c) wave-length. For this test series, HT-PV cells with the dimension of $10 \times 10 \times 0.5$ mm

were used. The results of open-circuit voltage and (I_{sc}) short-circuit (I_{sc}) current are shown in Fig. 2a and 2b.

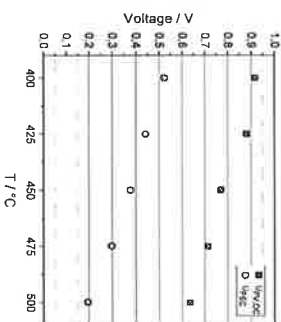


Fig. 2a. DC measurement of photo voltage on HT-PV and PFC cell at 400, 425, 450, 475 and 500°C under illumination with LED 10W/365 nm

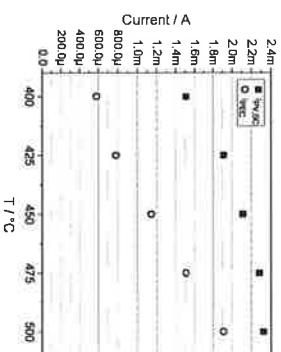


Fig. 2b. DC measurement of photo current on HT-PV and PFC cell at 400, 425, 450, 475 and 500°C under illumination with LED 10W/365 nm

The measurements have shown that U_{pvc} is decreasing with rising temperature, whereas I_{pvc} is increasing. In that context, it is interesting that the voltage at (2) the HT-PV cell pvc remains more or less stable over temperature. In the next step, the HT-PV cell was packed on top of an electrolyte supported cell (ESC) consisting of a polycrystalline 3 mol% yttria-stabilised zirconia (YSZ) substrate ($\varnothing 20 \times 0.3$ mm) coated on both sides with porous strontium doped lanthanum cobaltite (LSC) as electrode material for oxygen reaction. Voltage and current in the operation point (U_{pvc} , I_{pvc}) of the cell shown in Fig. 3 were measured at different temperatures. The measurements have shown that U_{pvc} decreases dramatically with rising temperature (Fig. 2a), whereas I_{pvc} increases (Fig. 2b).

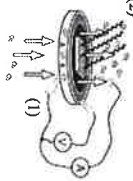


Fig. 3. Sketch of HT-PV on top of an ESC

Further on, the cell in Fig. 3 has been implemented in an experimental setup with gas-tight chambers and light-coupling by a quartz bar. When the light (LED 10W/365 nm) is switched on, oxygen is incorporated (Eq.1) at the cathode (1) due to the applied current and voltage from the HT-PV cell. Oxygen ions going through the YSZ following *Körger-Vink* nomenclature (Eq. 2) and leaves the cell at the anode (2), Eq. 3.

$$\frac{1}{2}O_2 + 2e^- \rightleftharpoons O^{2-} \quad \text{Eq. 1}$$

$$\frac{1}{2}O_2 + 2V_O^* \rightleftharpoons O_O^x + 2h^+ \quad \text{Eq. 2}$$

$$O^{2-} \rightleftharpoons \frac{1}{2}O_2 + 2e^- \quad \text{Eq. 3}$$

In a sequence of light off-and-on, voltage, current, and oxygen concentration have been measured at a temperature of 450°C, as shown in Fig. 4.

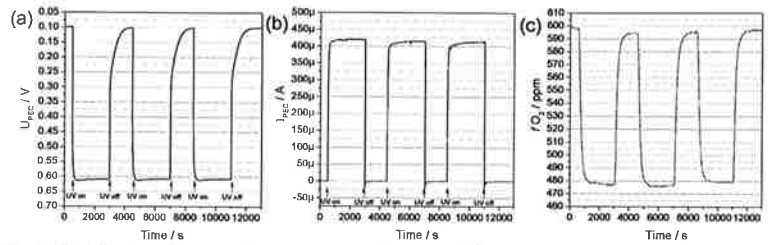


Fig. 4: Diagrams of a) voltage curve b) current curve and c) oxygen profile at 400°C

The voltage in the operation point under illumination has reached its maximum at nearly 0.5 V (Fig. 4.a). That has led to an electrical current of more than 0.8 mA (Fig. 4.b). Finally, the oxygen concentration has been shifted from 640 to 400 ppm O_2 (Fig. 4.c). Overall, it could be shown that the $La_{0.8}Sr_{0.2}CrO_3 / SrTiO_3(100)$ heterostructure is a promising material system for photo-electrochemical energy conversion.

Acknowledgement

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